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④ Heat-resistant chocolate-based confectionery products

⑤ A process for improving the heat resistance of chocolate or chocolate type products so that they have a reduced tendency to deform at elevated temperatures which comprises mixing a polyol gel or a polyol/water gel product in particulate form with a flowable mixture of chocolate type ingredients in an amount to achieve a polyol content of from 0.2 to 60% by weight based on the total amount of the resultant chocolate.

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The present invention relates to a process for improving the heat resistance of chocolate or chocolate type products so that they have a reduced tendency to deform or melt at elevated temperatures or to stick to the packaging material.

Chocolate products are usually thorough mixtures of liquid cocoa, cocoa butter, sugar, lecithin and possibly milk and flavouring substances. They therefore contain fatty substances which soften and melt between 30 and 35 °C.

When articles which consist entirely or partly of these products are exposed to temperatures lying above the melting point of the aforesaid fatty substances, (i.e. temperatures occurring during the summer season or in tropical countries), they tend to lose their original shape and appearance, and become soft and unpleasant to handle. If they are wrapped, the surface of the article may adhere to the wrapper and may produce unsightly fat staining.

In the case of chocolate type coatings intended to enrobe articles such as biscuits, confectionery, etc., the articles immediately lose their original appearance and are no longer suitable for consumption when they have been exposed to high temperature.

Different processes have been proposed in order to avoid the said disadvantages, but the products obtained, although more resistant to the effects of high temperature, have an unpleasant taste and a perceptibly coarser texture than that of normal conched chocolate.

In EP-B-189469 there is claimed a method of increasing the viscosity of a chocolate composition containing cocoa butter so that said composition will be substantially non-flowable at a temperature well above the normal melting point of cocoa butter characterised in comprising mixing 0.2 to 5 wt% of a liquid polyol selected from the group comprising glycerine, sorbitol, propylene glycol, mannitol, corn syrup and combinations and solutions thereof to a chocolate mixture after it has been tempered. However, a disadvantage of this process is that the chocolate tends to set rapidly e.g. from 40 to 60 seconds and this allows very little time for the further processing of the chocolate e.g. moulding into tablets of the desired shape or depositing a coating. A disadvantage of this method is that special additional equipment would be needed to incorporate the polyol at the latest possible time before further processing of the chocolate to prevent premature setting of the chocolate.

EP-A-459777 describes and claims a structured fat emulsion for incorporation into a confectionery coating as the fat ingredient thereof, said structured fat emulsion comprising:

30 a blend of a polyol component together with a fat component, said blend being a polyol-in-fat emulsion in which between about 2 and about 20 weight percent, based on the total weight of polyol-in-fat emulsion, of said polyol components is emulsified within said fat component, whereby said structured fat emulsion effects a time delay in the development of enhanced heat resistance and retardation of bloom development in a confectionery coating formulation when the formulation is subjected to setting conditions to form a set confectionery coating, and whereby these properties are substantially latent until the confectionery coating formulation is subjected to the setting conditions at least to the extent that the confectionery coating is flowable until subjected to the setting conditions. In this patent the polyol is blended separately with the fat ingredient of a confectionery coating before being added to the remaining ingredients of the confectionery coating so that the confectionery coating remains flowable until subjected to setting conditions. The confectionery coating may be a chocolate coating whereby the fat ingredient with which the polyol is blended may be, for instance, cocoa butter or a partially hydrogenated soybean fat. However, a disadvantage of a process of producing a confectionery coating containing a structured fat emulsion claimed in EP-A-459777 is that it cannot be used in the normal chocolate manufacturing process because the polyol has to be blended separately with the fat ingredient of the chocolate before adding to the remaining ingredients.

45 We have now found, surprisingly, that if a polyol gel or a polyol/water gel is added in particulate form to a flowable mixture of chocolate ingredients containing substantially the normal fat content for a conventional chocolate or chocolate coating, the resultant chocolate not only has improved heat resistance but the setting rate is similar to normal chocolate, i.e. very much slower than when using the method of EP-B-189469. This is because the polyol diffuses out of the gel gradually. In addition, this product can be used in the normal chocolate manufacturing process unlike the structured fat emulsion described in EP-A-459777. Hereinafter, the term "polyol gel" should be understood to mean either a polyol gel or a polyol/water gel.

50 Accordingly, the present invention provides a process for improving the heat resistance of chocolate or chocolate type products so that they have a reduced tendency to deform at elevated temperatures which comprises mixing a polyol gel product in particulate form with a flowable mixture of chocolate type ingredients in an amount to achieve a polyol content of from 0.2 to 60% by weight based on the total amount of the resultant chocolate.

55 The polyol gel may be formed by gelation of a polyol or a polyol/water mixture with a gelling agent. Examples of suitable gelling agents are gelatin, agarose, pectin, xanthan gum, agar agar, kappa-car-

rageenan, sodium alginate and iota-carrageenan. Other suitable gelling agents such as polysaccharides, e.g. glucomannan, may readily be selected by those skilled in the art. The gelling agents may be used individually or in any combination provided a suitably firm gel is formed with the polyol.

The polyol which is used for gelation is preferably a liquid and may be a straight or branched chain hydrocarbon based compound containing at least two hydroxyl groups on the carbon skeleton. For example, the polyol may be a dihydric alcohol, a trihydric alcohol such as glycerol, mannitol, sorbitol, propylene glycol or corn syrup or any combination thereof.

The polyol which is used for gelation may be a dry polyol containing no or substantially no water, for instance, less than 2%. The polyol/water mixture which is gelled may be a mixture containing up to 95% by weight of water, for example, from 5 to 85% and, if desired, from 10 to 50% by weight of water. When mixtures of polyol and water are used, the resulting chocolate may be less costly and have a lower calorie content than normal chocolate because a substantial portion of the chocolate's fat and sugar may be replaced by water and polyol. It may also be possible to incorporate water and/or polyol soluble aromas or flavours into chocolate.

We have found that the setting time or "hardening" time of the chocolate is increased when less water is used.

Hereinafter, the term "polyol" should be understood to mean either a dry polyol or a polyol/water mixture.

The gelation of the polyol may be carried out by a variety of gelation techniques depending on the gelling agent used. For example, ionic gelation is preferred when sodium alginate is the gelling agent. Thermal gelation is preferred in some cases, because it does not require the addition of substances other than the gelling agent.

Thermal gelation may be carried out, for instance, by slowly adding the gelling agent to the polyol with agitation below the gelling temperature to form a dispersion, raising the temperature to above the dissolution temperature, e.g. 90-170 °C, while maintaining the agitation and then cooling to below the gelling temperature e.g. ambient temperature. Gelling occurs between 80 ° and 50 °C. The amount of gelling agent used to form the gel may be from 0.5 to 15%, preferably from 1 to 7% and especially from 1.5 to 5% by weight based on the weight of the polyol.

If desired, a gum such as locust bean gum or gum arabic may be incorporated in the polyol gel product in which it may decrease syneresis and increase the break strength and resilience of the gel. Advantageously, the gum may be dispersed in the cold solution of the gelling agent in the polyol before gelation. The amount of gum may be up to 10%, preferably from 0.5 to 5% and especially from 1 to 3% by weight based on the weight of the polyol. The gum may conveniently be incorporated by dispersing in the cold dry polyol, increasing the temperature to dissolve the gum e.g. from 100 ° to 130 °C and cooling back to room temperature after which the gelling agent may be added to form the gel, e.g. according to the procedure described above. Other suitable gums will be readily determined by those skilled in the art.

Advantageously, certain food-grade alkali metal or alkaline earth metal salts, casein or cocoa powder may be added to the solution of the gelling agent in the polyol before gelation to alter the characteristics of the resultant gel. The alkali metal or alkaline earth metal salt may be for instance a sodium or calcium salt but is preferably a potassium salt e.g. KCl. The amount of alkali metal or alkaline earth metal salt added is preferably less than 1% by weight based on the weight of the polyol since amounts substantially greater than 1% may prevent gelation. The amount of casein e.g. as a potassium salt, or cocoa powder may vary more widely e.g. up to 20%, e.g. from 3 to 15% by weight based on the weight of polyol according to the choice. The addition of one or more of the above ingredients may contribute to the overall strength of the gel and this is particularly advantageous when the amount of gelling agent is low e.g. from 0.5 to 1.5% by weight. Low amounts of gelling agent may be used if short "hardening" or "setting" times are required for the chocolate.

The polyol gel should preferably disperse uniformly into the chocolate mass and this is achieved by forming the gel into particles, especially beads.

One method of forming a particulated polyol gel involves freezing the prepared polyol gel e.g. to a temperature of from about -170 °C to -200 °C and then grinding in a cold grinder. The mean diameter of the particles or beads may be from 1 to 1000 microns, preferably from 5 to 400 microns and especially from 10 to 100 microns. Advantageously, an anti-caking agent such as cocoa powder may be added to the particulate cold polyol gel and the mixture heated to room temperature whereupon it turns into a free flowing powder since cocoa particles prevent the gel beads from agglomerating. During the addition of the cocoa powder to the polyol gel care should be taken to prevent moisture from the environment being drawn into the mixture. The amount of polyol gel in the mixture is preferably from 10 to 80% by weight, and especially from 40 to 70% by weight.

A second method of forming a particulated polyol gel comprises adding a cold polyol solution containing dispersed gelling agent to cocoa butter above its melting point, heating to dissolve the gelling agent e.g. from 90° to 170°C with agitation e.g. stirring, to disperse the polyol droplets (containing gelling agent) throughout the cocoa butter phase to form an emulsion. The droplet size may be reduced by increased agitation and for particle sizes below about 100 microns, a homogeniser may be used. If desired, an emulsifier such as lecithin may be added. After the required droplet size is obtained, the mixture may be allowed to cool, e.g. to room temperature, while continuing the agitation to maintain the droplets uniformly dispersed in the cocoa butter matrix. When the temperature falls below the gelling point, the droplets turn into gel beads. When the temperature falls below the melting point of the cocoa butter (30-38°C), the viscosity of the mixture increases, and finally the mixture solidifies and prevents coalescence of the gel beads. If desired, an excess of cocoa butter may be separated from the gel beads while the mixture is still in the liquid phase e.g. by decantation, centrifugation or filtration.

The polyol gel product may be added to the flowable chocolate or chocolate type ingredients at any suitable stage e.g., in the case of chocolate and certain chocolate type products, before, during or after tempering as long as the polyol gel remains in solid particulate form during the mixing and during subsequent processing which may include tempering and afterwards moulding into tablets or bars of the desired shape, or depositing a coating. The polyol gel is preferably in bead or powdered form.

The flowable mixture of chocolate or chocolate type ingredients may consist of the ingredients used for conventional plain milk or white chocolate, substitute or imitation chocolate or low calorie chocolate to be used for chocolate bars or for coatings.

The amount of polyol in the resultant chocolate may be from 0.5 to 10%, preferably from 0.75 to 5% and most preferably from 1 to 2% by weight based on the total weight of the chocolate.

The process of this invention may be used for preparation of low cost chocolate or chocolate products (because high cost fat can be replaced with water and/or glycerin), for low calorie chocolate or chocolate products (because of replacement of fat by water), and for the addition of water or glycerin soluble aromas and flavors into chocolate or chocolate products.

The following Examples further illustrate the present invention. Parts and percentages are given by weight.

30 Example 1

- a) 100 parts of dry glycerin were continuously agitated by a magnetic stirrer and 3 parts of kappa-carrageenan were added slowly at a temperature below the gelation temperature in order to avoid incipient gelation and lump formation. The temperature was raised to 120°C while maintaining the stirring during which time the carrageenan was first in dispersion and as the temperature increased it entered into solution. The solution was then allowed to cool to ambient temperature during which time gelation occurred between 80° and 50°C.
- b) The gel was then frozen in liquid nitrogen to -185°C and ground to a particle size of 200 microns by a mortar and pestle in a cold grinder. 100 parts of cocoa powder were added and the temperature allowed to rise to room temperature while ensuring that moisture from the environment was not absorbed into the mixture. At room temperature, the mixture turned into a free flowing powder in which the cocoa particles prevent the gel beads from agglomerating.
- c) The glycerin gel prepared above was then used to prepare heat-resistant chocolate by the following procedure:

45 100 parts of milk chocolate were melted in an oven at 40°C. 2 parts of the glycerin gel prepared above were mixed into the melted chocolate and thoroughly dispersed using a spatula to give a chocolate containing 1% by weight of glycerin.

In order to determine the period of time before the chocolate sets, the following testing procedure was used:

50 Commercially available milk chocolate was melted at 40°C. The desired amount of glycerin gel was added to the molten chocolate and continuous stirring was applied by hand for about five minutes while the temperature of the chocolate was above its melting point. If during this time no considerable increase of viscosity was observed, it was assumed that there was sufficient retention of glycerin in the gel structure. Subsequently, the chocolate was allowed to set at room temperature and it was stored at ambient conditions. During the next several days, from time to time the chocolate sample was placed in an oven at 40°C. If the sample became soft, it was taken out of the oven and stored at room temperature for a longer period of time. On the other hand, if the sample did not soften, that was an indication that glycerin had been released from the gel matrix and hardened the chocolate. The time required for this is defined as

"hardening time".

In this case, the chocolate containing the glycerin gel had not reached its hardening time even after 50 days.

5 **Examples 2 to 5**

A similar procedure to that described in Example 1a for preparing a glycerin gel was followed, but adding to the glycerin, in addition to 3 parts of kappa-carrageenan, 3 parts of locust bean gum, 1 part of potassium caseinate and 0.2 parts of KCl. The gel was frozen and ground as in Example 1b and cocoa powder was added to the frozen gel in the ratios shown in Table 1, and the resultant glycerin gel added to 100 parts of milk chocolate in the amounts shown in Table 1. The percentage amount of glycerin added to the chocolate and the time taken for the chocolate to harden are also indicated in Table 1.

TABLE I

EXAMPLE	GEL/POWDER WEIGHT RATIO	AMOUNT ADDED	% GLYCERIN IN CHOCOLATE	HARDENING
20 2	40:60	0.5	2	SOFT AFTER 40 DAYS
3	60:40	0.83	5	2 DAYS
4	70:30	0.285	2	SOFT AFTER 40 DAYS
25 5	60:40	0.67	4	3 DAYS

COMPARATIVE EXAMPLE

30 2 parts of free glycerin were added to 100 parts of milk chocolate melted at 40 °C. After stirring for only 30 seconds, the mix became very hard and further mixing was impossible.

Example 6

35 3 parts of kappa-carrageenan, 3 parts of locust bean gum, 1 part of potassium caseinate and 0.2 parts of KCl were added in 100 parts of dry glycerin and were dispersed at a temperature below the gelation temperature.

40 40 parts of the glycerin gel was added to 100 parts of liquid Malaysian cocoa butter and heated to 100 °C. The glycerin droplets were dispersed into the butter phase by means of a homogeniser and, afterwards, the mixture was cooled to give a butter phase containing gel beads having a mean diameter of 200 microns.

45 5 parts of the butter/glycerin gel as prepared above were mixed into 100 parts of milk chocolate melted in an oven at 40 °C and thoroughly dispersed using a spatula to give chocolate containing 2% by weight of glycerin. By following the test procedure described in Example 1 the chocolate containing the glycerin gel hardened after 40 days.

50 It is apparent from the results described in the foregoing Examples that the carrageenan gels can be used to control the release of glycerin into the chocolate mass. Therefore, it is possible to process chocolate that contains glycerin without the undesired increase in viscosity. Upon storage, glycerin will release, gradually, thus imparting heat resistance to the resultant chocolate.

Examples 7-16

A similar procedure to that described in Example 6 for preparing a glycerin gel was followed but using, instead of kappa-carrageenan, the following gelling agents which are available commercially from FMC Corporation, Food Ingredients Division, 1735 Market St, Philadelphia, PA 19103, USA.

Gelcarin DG654B (a mixture of kappa-carrageenan and locust bean gum)

Gelcarin DG556B (a mixture of iota-carrageenan and locust bean gum)

Gelcarin GP359 (kappa-carrageenan)

Gelcerin GP812 (kappa-carrageenan with potassium and sodium ions)

Experiments were performed in order to study the effects of the type of gelling agent, concentration of the gelling agent, gel bead size and the amount of gel added to the chocolate on the effective "hardening" time of chocolate samples. The results are shown in the following Table II where the measurements are only 5 qualitative and should be considered only an approximation. The hardening times are also approximate since the samples were not monitored every day.

TABLE II

10	Example	Gelling Agent	Conc. % gelling agent	Bead size (μ)	Gel % added to chocolate	Hardening Time (days)
15	7	Gelcerin DG654B	2	< 100	3.3	8
20	8	Gelcerin DG654B	6	< 100	7.2	9
25	9	Gelcerin DG654B	6	> 300	7.6	29
30	10	Gelcerin DG556B	3	< 100	10.6	13
	11	Gelcerin DG556B	6	> 300	6.1	21
	12	Gelcerin GP812	2	< 100	7.0	19
	13	Gelcerin GP812	4	< 100	6.8	16
	14	Gelcerin GP812	6	< 100	10.4	16
	15	Gelcerin GP812	6	> 300	6.7	29
	16	Gelcerin GP359	2	< 100	6.0	12

Effect of Type of Gelling Agent on Hardening Time

35 It can be seen that the shortest hardening times (8-12 days) are achieved by DG654B and by FP359, which are the kappa-carrageenan agents, and by the small gel beads. Medium hardening times (13 days) are achieved by DG556B which is the iota-carrageenan agent. Longer hardening times (16-19 days) are required for GP812, which is kappa-carrageenan combined with potassium and sodium salts.

40 Effect of Gelling Agent Concentration and Amount of Gel Added to Chocolate on Hardening Time

There is a slight tendency, if any, for the hardening time to increase with increasing concentration of the gelling agent, as is suggested from Examples 7 to 11. More profound is the effect that the amount of gel added to the chocolate has on the hardening time.

45 By comparing Examples 14 and 15 where the concentrations of the gelling agent are the same (6%), the hardening time was increased from 16 to 29 days when the amount of glycerin gel added to the chocolate was reduced from 10.4% to 6.7%.

Effect of Gel Bead Size on Hardening Time

50 It is apparent that for each gelling agent used, the hardening time is longer when the gel bead size is larger. So, for DG654B the hardening time is 8 days for the small gel beads and 29 days for the large. For DG556B, the hardening time is 13 days for the small gel beads and 21 for the large beads. Finally, for GP812, the hardening time is 16-19 days for the small gel beads and 29 days for the larger gel beads.

Examples 17-23

A similar procedure to that described in Example 6 for preparing a glycerin gel was followed but using, instead of kappa-carrageenan, Gelcarin DG654B at a concentration of 6% based on the gel and, instead of dry glycerin, a mixture of glycerin and water as indicated in Table III below:

TABLE III

Example	Water-to-glycerol ratio	Bead size (μ)	% gel added to chocolate	Hardening Time (days)
17	50/50	> 300	9.6	7
18	60/40	> 300	6.4	5
19	70/30	> 300	8.1	5
20	80/20	> 300	2.5	5
21	40/60	< 100	6.7	3
22	50/50	< 100	5.6	3
23	60/40	< 100	4.3	3

Experiments were performed in order to study the effects of the gel bead size, and water content of the gel on the effective "hardening" time of chocolate samples. The measurements are only qualitative and should be considered only an approximation. The hardening times are also approximate since the samples were not monitored every day.

Table III shows that the addition of water/glycerol gels results in much shorter hardening times of chocolate (compare to Table II that lists the hardening times for various glycerol gels). This is possibly due to the small size of the water molecules, which permits them to exude from the gel structure. It is also apparent from Table II that the smaller the gel bead size, the shorter the hardening time.

Claims

1. A process for improving the heat resistance of chocolate or chocolate type products so that they have a reduced tendency to deform at elevated temperatures which comprises mixing a polyol gel product in particulate form with a flowable mixture of chocolate type ingredients in an amount to achieve a polyol content of from 0.2 to 60% by weight based on the total amount of the resultant chocolate.
2. A process according to claim 1 wherein the gelation is effected by thermal gelation.
3. A process according to claim 1 wherein the amount of gelling agent used to form the gel is from 0.5 to 15% by weight based on the weight of the polyol.
4. A process according to claim 1 wherein the polyol used is in dry or substantially dry form.
5. A process according to claim 1 wherein the polyol may be used in admixture with up to 95% by weight of water.
6. A process according to claim 1 wherein a gum, a food grade alkali metal or alkaline earth metal salt, casein or cocoa powder or any mixture thereof are dispersed in a cold solution of the gelling agent and polyol before gelation.
7. A process according to claim 1 wherein the mean diameter of the particles of the polyol gel product is from 1 to 1000 microns.
8. A process according to claim 1 wherein the polyol gel is frozen and then ground in a cold grinder, cocoa powder is added to the particulate cold polyol gel and the mixture heated to room temperature to form a free flowing powder.

9. A process according to claim 1 wherein a cold polyol dispersion containing a gelling agent is added to melted cocoa butter, heating the mixture with agitation to dissolve the gelling agent and then cooling to a temperature below the gelling temperature to form gel beads suspended in the butter phase and finally, separating the gel beads from excess butter phase.
- 5 10. A process according to claim 1 wherein the polyol gel product is mixed with a flowable mixture of chocolate type ingredients before, during or after tempering of the chocolate or chocolate type ingredients.
- 10 11. Chocolate or chocolate type products whenever produced by a process according to any of the preceding claims.

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EUROPEAN SEARCH REPORT

Application Number
EP 95 20 1077

DOCUMENTS CONSIDERED TO BE RELEVANT															
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)												
P, X	EP-A-0 637 420 (NESTLE) 8 February 1995 * the whole document * ---	1-4, 7, 10, 11	A23G1/00												
X	EP-A-0 437 360 (WARNER-LAMBERT) 17 July 1991 * page 3, line 3 - page 7, line 24; claims 1-5, 15, 19 * ---	1, 2, 7, 10, 11													
X	WO-A-93 19613 (MARS) 14 October 1993 * page 11, line 10-19; claims 1-4, 16; example 5 1 * ---	1													
X	WO-A-91 19424 (MARS) 26 December 1991 * page 2, paragraph 5 - page 3, paragraph 1; claims; examples 1, 11 * ---	1, 2, 5													
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X	WO-A-93 06737 (RAFFINERIE TIRLEMontoise) 15 April 1993 * page 6, line 38 - page 10, line 24; claims 1-7, 10-13; examples 1-6 * ---	1	TECHNICAL FIELDS SEARCHED (Int.Cl.6)												
A	GB-A-2 065 442 (MORINAGA & CO.) 1 July 1981 * claim 1; examples * ---	1, 8	A23G												
A	WO-A-89 12403 (THE NUTRASWEET COMP.) 28 December 1989 * the whole document * ---	1													
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<p>The present search report has been drawn up for all claims</p> <table border="1"> <tr> <td>Place of search</td> <td>Date of completion of the search</td> <td>Examiner</td> </tr> <tr> <td>THE HAGUE</td> <td>28 August 1995</td> <td>Guyon, R</td> </tr> <tr> <td colspan="2">CATEGORY OF CITED DOCUMENTS</td> <td> T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application I : document cited for other reasons A : member of the same patent family, corresponding document </td> </tr> <tr> <td colspan="2"> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document </td> <td></td> </tr> </table>				Place of search	Date of completion of the search	Examiner	THE HAGUE	28 August 1995	Guyon, R	CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application I : document cited for other reasons A : member of the same patent family, corresponding document	X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		
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DOCUMENTS CONSIDERED TO BE RELEVANT									
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A,D	EP-A-0 459 777 (BUNGE FOODS CORP.) 4 December 1991 * column 4, line 8-43 * * column 5, line 8-18; example 1 *	1							
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A	DATABASE WPI Week 9119 Derwent Publications Ltd., London, GB; AN 91-136288 & JP-A-03 072 846 (KYOWA HAKKO KOGYO) , 28 March 1991 ---								
A	US-A-2 760 867 (N. W. KEMPF ET AL.) 28 August 1956 * abstract *	1	TECHNICAL FIELDS SEARCHED (Int.Cl.6)						
<p>The present search report has been drawn up for all claims</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 33%;">Place of search</td> <td style="width: 33%;">Date of completion of the search</td> <td style="width: 34%;">Examiner</td> </tr> <tr> <td>THE HAGUE</td> <td>28 August 1995</td> <td>Guyon, R</td> </tr> </table> <p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons R : member of the same patent family, corresponding document</p>				Place of search	Date of completion of the search	Examiner	THE HAGUE	28 August 1995	Guyon, R
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